

PC - 376 CV-19  
(533) M.Sc. Chemistry (Third Semester)  
Examination Dec-2020  
Compulsory/Optional  
Physical Organic Chemistry  
Group -  
Paper -

Time : Three Hours]

[Maximum Marks : 80  
[Minimum Pass Marks :

नोट : दोनों खण्डों से निर्देशानुसार उत्तर दीजिए। प्रश्नों के अंक उनके दाहिनी ओर अंकित हैं।  
Note : Answer from both the Section as directed. The figures in the right-hand margin indicate marks.

Section-A

1. Answer the following questions in few words:
- (a) Which halogen nucleophile is weakest in polar aprotic solvents?  
(i)  $I^-$  (ii)  $F^-$  (iii)  $Cl^-$  (iv)  $Br^-$
- (b) Give example of typical Hard Acids.
- (c) Which of the following compounds would be the best nucleophile.  
(i)  $CH_3S^-$  (ii)  $CH_3O^-$  (iii)  $NH_3$  (iv)  $CH_3SH$
- (d) Rank the following compounds in order of increasing nucleophilicity.  
(i)  $NH_3$  (ii)  $H_2O$  (iii)  $Cl^-$  (iv)  $F^-$
- (e) What is Curtin - Hammett principle.
- (f) What do you understand by steric strain in organic compound?
- (g) What is the difference between hyper conjugation and resonance?
- (h) State whether the following is aromatic or antiaromatic or non-aromatic.

1 X 10



- (i) What three critical factors affect Gibbs free energy change.
- (j) Carbon monoxide has ten bonding electrons and four antibonding electrons. Calculate the bond order.

2 X 5

2. Answer the following questions:
- (a) What do you understand by aromaticity?
- (b) Explain ambivalent - nucleophile with example.
- (c) What is Marcus theory of electron - transfer.
- (d) What is Huckel rule? Explain with example.
- (e) Explain nucleophilicity scale.

Section-B

Unit-I

3. Explain Huckel molecular orbital method as a means to explain modern theoretical methods.

12

Or

Write notes on : (a) Conjugation and hyper conjugation (b) Orbital Symmetry

Unit-II

4. What are hard and soft acids and bases : Explain acidity functions and their application.

12

Or

Explain the following : (a) Acid - base catalysis (b) Micellar Catalysis

**Unit-III**

12

5. Explain the various types of steric strain and their influence on reactivity.

**Or**

Write notes on : (a) Hammond's Postulate (b) Radical Stability

**Unit-IV**

12

6. Discuss in detail the electron transfer nature of  $SN_2$  reactions.

**Or**

Explain in detail :

- (a) Structural effects on rates and selectivity.  
(b) Curve crossing approach to electrophilic reactivity.

**Unit-V**

12

6. Explain various types of steric strain and their influence on reactivity.

**Or**

Write notes on : (a) Winstein – Holness and Curtin Hammett Principles (b) Steric LFER